

Operating Instructions

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UME^x**300** Passive Sampler for Ammonia Cat. No. 500-300

The SKC UME^x 300 Passive Sampler effectively collects ammonia at 39.92 ml/min. The UME^x 300 Passive Sampler contains a tape treated with sulfuric acid. Each sampler incorporates a "blank/correction" section of tape in addition to the sample section of tape. Analysis is by ion chromatography (IC) with conductivity detection for identification of ammonia that may be present in the sample. Designed for single use, the SKC UME^x 300 Passive Sampler is packaged in an aluminized pouch for easy sample transport to a laboratory.

Performance Profile

Sampling Rate: 39.92 ml/min with a relative standard deviation of 11.4%

Validated

Concentration Range: 0.1 to 100 ppm

Analysis Method: Solvent extraction and analysis of ammonium ion by

IC with conductivity detection

Lower Detection

Limits:

15 minutes: 2.4 ppm 8 hours: 0.075 ppm

24 hours: 0.025 ppm

Shelf-life: 9 months from date of manufacture in freezer at

 \leq 39.2 F (4 C)

Storage: Before use: Freezer at \leq 39.2 F (4 C)

After use: Analyze within 3 weeks. Store at freezer $(\leq 39.2 \text{ F } [4 \text{ C}])$ or ambient (72 F [22 C]) temperatures

Accuracy: $\pm 23.4\%$

Temperature Effects: No effect on sampling rate from 22 to 40 C

Humidity Effects: No effect from 20 to 80% relative humidity (RH)

Wind Velocity Effects: No effect from 5 to 100 cm/sec

Interferences: None identified

Dimensions: $3.4 \times 1.1 \times 0.35 \text{ in } (8.6 \times 2.8 \times 0.89 \text{ cm})$

Weight: 0.38 oz (10.9 gm)

Slide Cover: Pink

Sampling Instructions

Cautions: • Do not store with food.

- Before sampling, check the expiration date on the label on the outside of the pouch. Do not use after the last day of the month indicated.
- SKC recommends using gloves when handling chemically treated media.
- UME^x samplers are designed for single use only. Do NOT reuse UME^x samplers.
- Open the aluminized pouch and remove the sampler. Do not discard the pouch; use it to send sampler to the laboratory. Store the pouch away from potential ammonia sources.
- 2. Enter date, start time, and location in the space provided on the back of the sampler.
- 3. Position the sampler on a worker's collar for personal sampling or in an appropriate location for area sampling.
- 4. Slide the sampler cover to the "on" position to begin sampling.
- 5. After sampling for the desired time, up to 24 hours, slide the sampler cover to the "off" position to stop sampling. Enter the sample stop time in the space provided on the back of the sampler.
- 6. Place the sampler in the original pouch immediately after sampling.
- 7. Seal the pouch. Send pertinent information and sample to an accredited laboratory for analysis.

Analysis Instructions for Ammonia Calibration Standards – Ammonium

Purchase commercially available ammonium standards appropriate for your application. Prepare a range of ammonium standards in de-ionized ultra-filtered (DIUF) water.

Note: The ammonium curve is not linear. A graphing program will be needed to calculate results.

Sample Preparation for Ammonia

- 1. Remove the sampler from the pouch and the sliding cover from the sampler.
- 2. Use clean forceps to lift out the treated tape from each section. Place each piece of tape in a sealed vial. This provides a sample and a blank/correction.

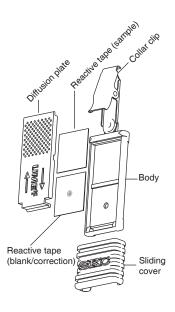
Note: The blank/correction section of tape has an indentation for easy identification.

- 3. Ammonium is desorbed from both the sample and the blank/correction tapes by adding 3 ml of DIUF water to each vial. Shake the sample by hand and then let stand for 60 minutes. Excessive shaking causes the filter to break apart in the vial.
- Immediately transfer 1 ml of extract to an auto-sampler vial for analysis of ammonia.

Ammonia Sample Analysis

- 1. The sample extracts are analyzed for the ammonium ion by ion chromatography with conductivity detection.
- 2. A 25 microliter portion of the extract is injected onto a Dionex 3 x 250-mm CS16 column and with a 36 mM methanesulfonic acid eluent.
- Calculate the ammonium results by comparing against a standard calibration curve.
- 4. Convert the results from ammonium to ammonia using the following formula: Concentration μ g/ml ammonia = Concentration μ g/ml ammonium x 0.944
- 5. Total mass of ammonia is calculated below: Concentration ammonia (µg/ml) x Desorption Volume (3 ml)
- 6. The ammonia from the blank/correction tape must always be subtracted from the sample tape when calculating air concentrations.
- 7. Calculate the air concentration in ppm using the following equations:

Volume of air (liters) = $\frac{\text{Time (minutes) x Sampling rate (39.92 ml/min)}}{1000}$ Concentration (ppm) = $\frac{\text{Mass (mg) x 24450}}{\text{Air Volume (L) x Molecular weight (17.0 g/mole)}}$



UME ^x Passive Samplers [#]	Cat. No.
UME ^x 300,*† for ammonia, pk/10	500-300
UME ^x 100,* ^{†‡} for formaldehyde and other aldehydes, pk/10	500-100
UME ^x 200,*† for sulfur dioxide and/or nitrogen dioxide, pk/10	500-200
Accessory	
Treated Tape, for QC purposes only, pk/25	P20083
Stand for Indoor Sampling	690-302

^{*} Limited shelf-life

SKC Limited Warranty and Return Policy

SKC products are subject to the SKC Limited Warranty and Return Policy, which provides SKC's sole liability and the buyer's exclusive remedy. To view the complete SKC Limited Warranty and Return Policy, go to http://www.skcinc.com/warranty.asp.

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[†] Do not store with food.

[‡] Storage in a freezer at $\leq 39.2 F(4 C)$ required

[#] UME^x passive samplers are designed for single use only. Do NOT reuse UME^x samplers.